A VERSATILE NEW SYNTHESIS OF QUINOLINES AND RELATED FUSED PYRIDINES. PART IV. A SIMPLE ONE-POT ROUTE TO PYRIDO[2,3-5] QUINOLIN-2-ONES FROM ANILIDES.

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<u>Abstract</u>: The title pyridoquinolin-2-ones are readily made by sequential treatment of an acetanilide firstly with DMF and POCL<sub>2</sub> and then with another secondary amide.

In earlier papers in this series we have outlined a versatile approach to 2-chloro-3-formylquinolines (1) from acetanilides under Vilsmeier conditions.<sup>2</sup> A minor by-product from

acetanilides under these conditions (readily observed by its bright blue fluorescence under u.v. during t.l.c. examination) was found to be the pyridoquinolinone (2; R=R, R'=Ph), formally derivable from (1) and acetanilide (dotted lines in 2). Although reaction of (1) with acetanilide under a variety of conditions failed to give any (2), the latter product formed in reasonable yield (25%) when 2 moles of acetanilide was reacted with DMF (2M) and POCl<sub>3</sub> (7M) at 75° for 18 h, suggesting that the protected chloro-aldehyde (1) indeed reacts with a derivative of acetanilide, probably the bidentate nucleophilic enamine (3) (Scheme). We have demonstrated already that this enamine is a key intermediate in the conversion of acetanilide to 2-chloroquinoline-3-aldehyde. 2h

The pyridoquinolinones (2) are best made as follows: A 2-chloroquinoline-3-aldehyde (1)

$$e.g. CHXY = CH NMe2 POC13 POC13 POC13 POC13 Ph Scheme$$

is prepared as described previously<sup>2</sup> by the action of DMF (2M) and POCl<sub>3</sub> (7M) on an acetanilide (1M). To this reaction mixture is added a secondary amide (1M) in POCl<sub>3</sub> (7M) solution containing one drop of DMF and the mixture heated for a further 2-3 h at 75°. The mixture is poured into ice-water, make alkaline (pH 9) and the crude product extracted and triturated with ice-cold methanol. By this means a variety of pyridoquinolinones have been generated in a simple one-pot procedure.

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Table

Starting Acetanilide	Derived Quinoline (1) R=	Added Amide	Product (2)		Yield	M.p.
			R	R*	(%)	(°c)
PhNHAc	н	PhNHAc	н	Ph	59	256-6
m-MeC <sub>6</sub> H <sub>4</sub> NHAc	7-Me	p-MeOC <sub>6</sub> H <sub>4</sub> NHAc	8-Me	р-ме <sup>ОС</sup> 6 <sup>Н</sup> 4	33	270-2
m-MeC <sub>6</sub> H <sub>4</sub> NHAc	7-Me	p-C1C <sub>6</sub> H <sub>4</sub> NHAc	8-Me	p-C1C <sub>6</sub> H <sub>4</sub>	25	300-1
m-MeC <sub>6</sub> H <sub>4</sub> NHAc	7 <b>–</b> Me	EtNHAC	8-Me	Et	22	169-172
m-MeC <sub>6</sub> H <sub>4</sub> NHAc	7-Me	PhCH <sub>2</sub> NHAc	8-Me	PhCH <sub>2</sub>	26	210-11
PhNHAc	H	PhNHCOEt	3-He	Ph	27	244-5

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## References

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